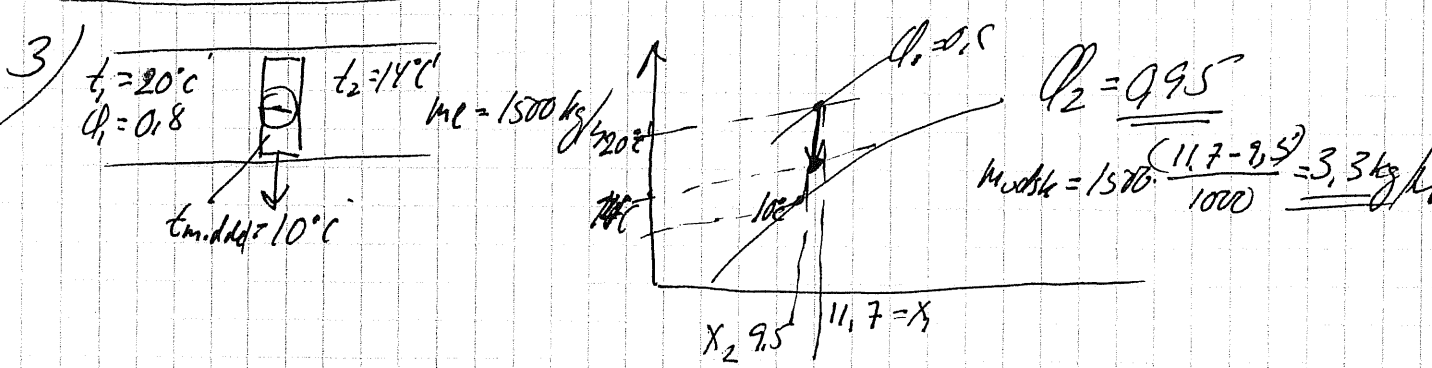
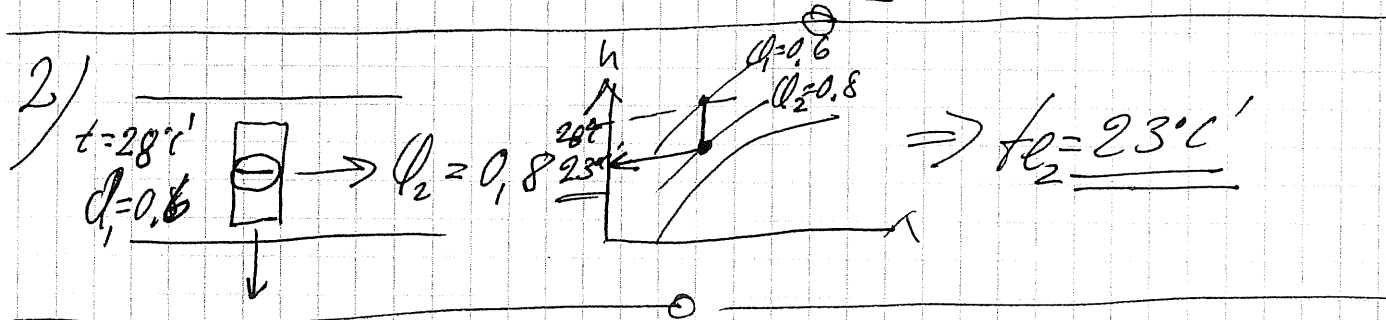


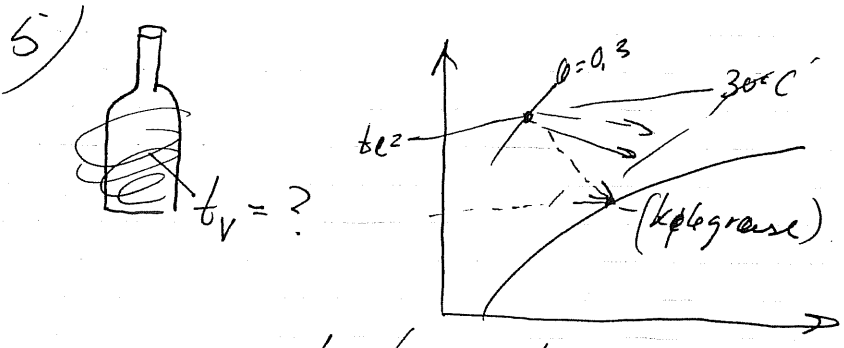
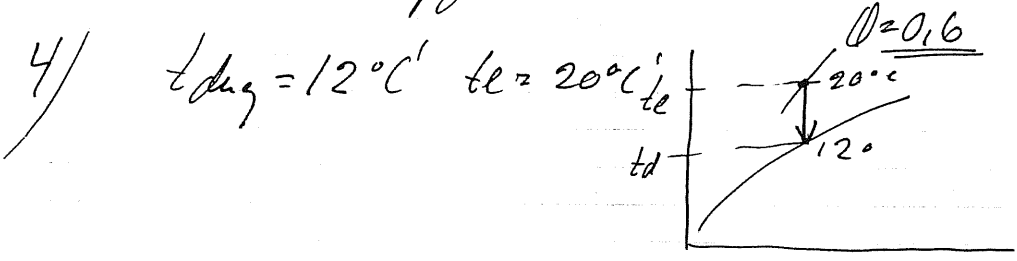
Øvelsesopgaver i Ventilations teknik

- 1:
- $x = \frac{m_{damp}}{m_{luft}}$
 - $h = 2500x + (2x+1)t$ [kJ/kg tør luft]
 - $\phi = \frac{m_{damp}}{m_{mættet damp}}$
 - $20^\circ C \Rightarrow t_{ind} = 23 \text{ mbar}$
 - $\left. \begin{matrix} t_g = 20^\circ C \\ \phi = 0.8 \end{matrix} \right\} \Rightarrow h = 50 \text{ kJ/kg}$
 - $x = 11.7 \text{ g/kg}$ og $p_{id} = 18 \text{ mbar}$
 - dugpunkt = den laveste temperatur, som luften kan afkøles til, uden at udskille vand
 - kølegrense: Den laveste temperatur omv et vædt legeme kan afkøles til, med luft af en given fugtighed.
 - $\left. \begin{matrix} t_e = 20^\circ C \\ \phi = 0.6 \end{matrix} \right\} \Rightarrow \left. \begin{matrix} t_{dug} = 11.5^\circ C \\ t_{køl} = 15.5^\circ C \end{matrix} \right\}$

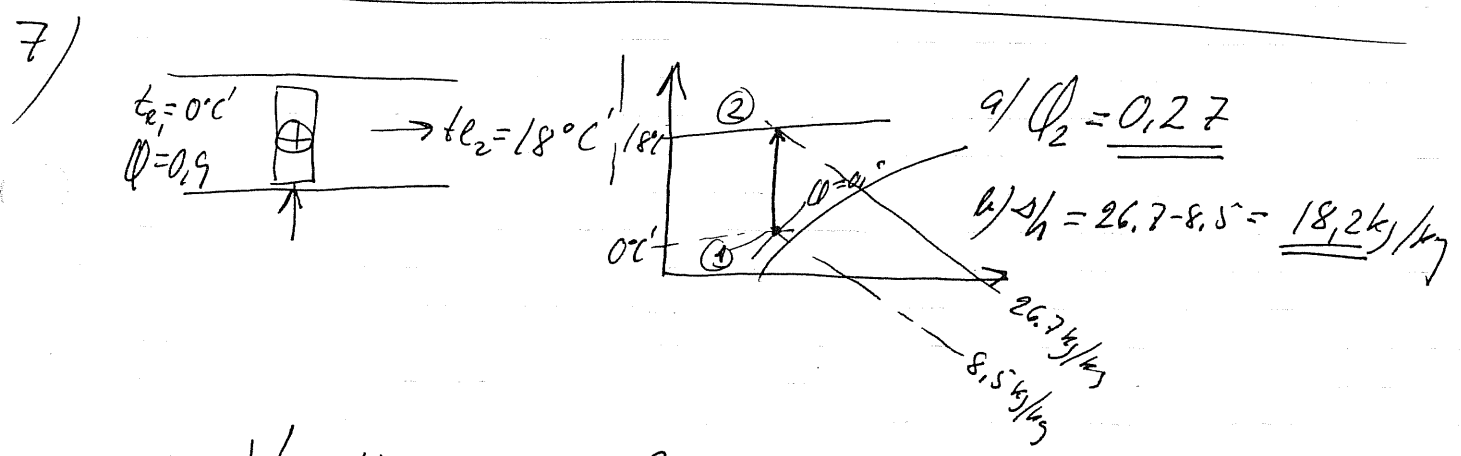
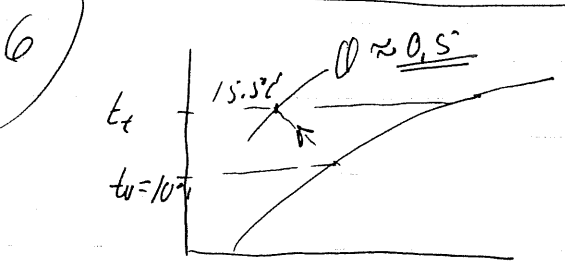


Luft kølegrens endelvirkningsgrad = $\frac{50 - 38}{50 - 29.5} = 0.585$

- " - temperatur virkningsgrad = $\eta_t = \frac{20 - 14}{20 - 10} = 0.6$



- a) $t_{\min} = t_{\text{kølegrænse}} = \underline{17.5^\circ\text{C}}$
- b) Øllets temperatur vil stige



$V_{\text{rum}} = 4 \times 8 \times 3 = 96 \text{ m}^3$

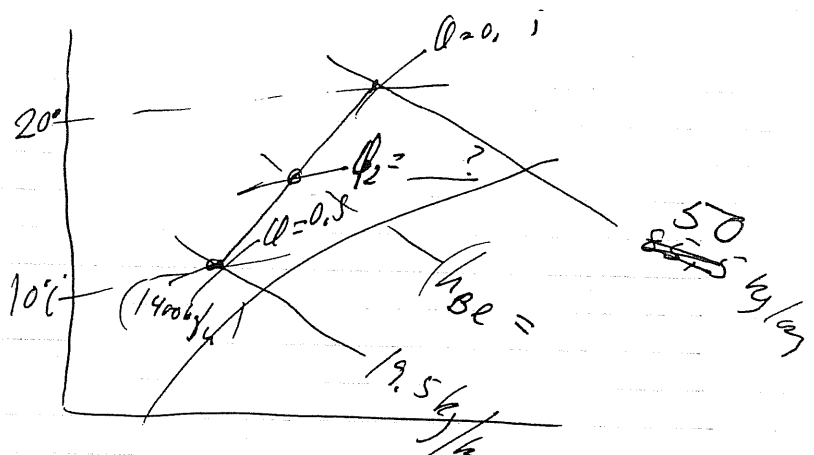
$V_{\text{ved}} = 4 \times 96 = \underline{384 \text{ m}^3/\text{h}}$

$m_{\text{ved}} = 384 \times 1.293 = \underline{500 \text{ kg/h}}$

$P_{\text{delf}} = \frac{m_{\text{ved}} \cdot s_h}{3600} = \frac{500 \cdot 18.2}{3600} = \underline{2.5 \text{ kW}}$

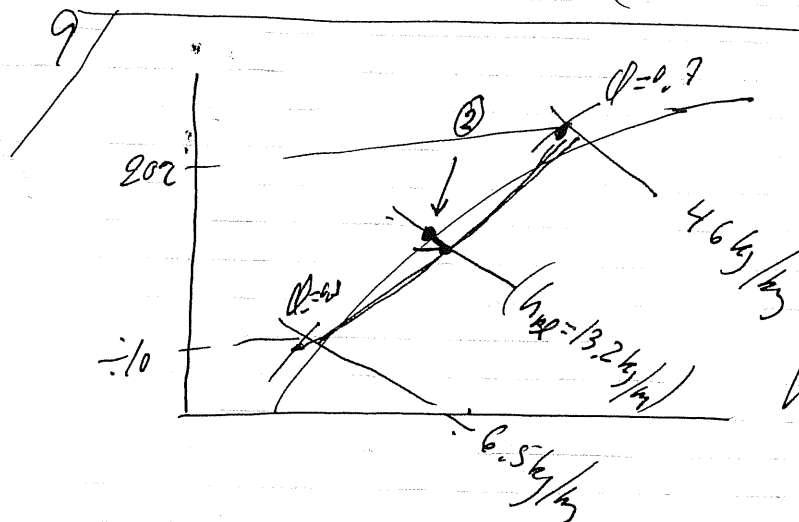
(NB: der ses brud for luftens fugtigheds)

$t_1 = 10^\circ\text{C}$
 $Q = 0.15$
 $m_2 = 600 \text{ kg/h}$
 $t_2 = 20^\circ\text{C}$
 $Q = 0.8$



$$h_{BL} = \frac{1400 \cdot 19.5 + 600 \cdot 50}{1400 + 600} = \underline{\underline{28.7 \text{ kg/kg}}}$$

$$Q_2 = \underline{\underline{0.65}}$$



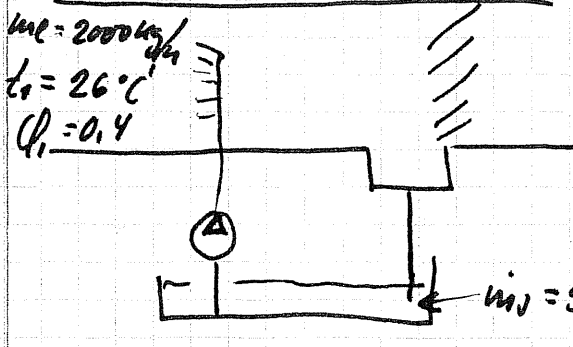
$$h_{BL} = \frac{500 \cdot 6.5 + 300 \cdot 46}{800} = \underline{\underline{13.2 \text{ kg/kg}}}$$

(ligger i kogrområdet)

$$\Rightarrow \underline{\underline{Q = 1}}$$

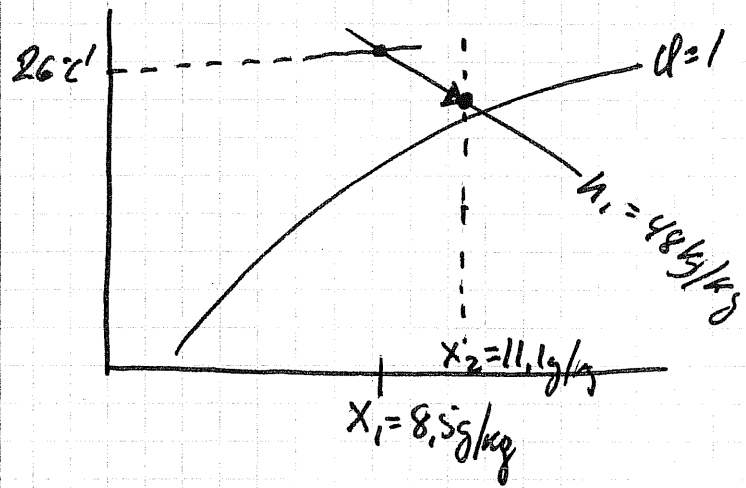
$$\Delta X = 4.7 - 4.3 = \underline{\underline{0.4 \text{ kg/kg}}}$$

10



$$\Rightarrow \Delta X = \frac{m_w}{m_e} = \frac{5.2 \cdot 10^3}{2000} = \underline{2.6 \text{ g/kg}}$$

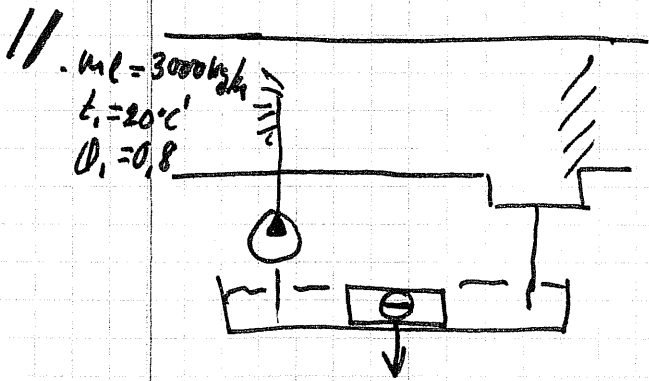
$$X_2 = X_1 + \Delta X = 8.5 + 2.6 = \underline{11.1 \text{ g/kg}}$$



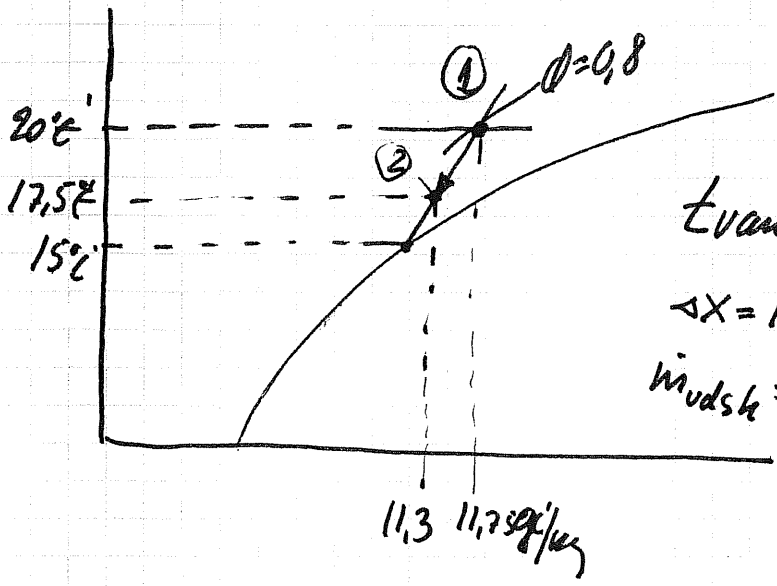
$$\Rightarrow \phi_2 = \underline{0.77} \text{ (after evaporation)}$$

$$t_2 = \underline{19.5^\circ\text{C}} \text{ (---)}$$

$$\Delta X = \frac{11.1 - 8.5}{12.25 - 8.5} \approx \underline{0.7}$$



$t_2 = 17.5^\circ\text{C}$
 $\phi_2 = 0.9$



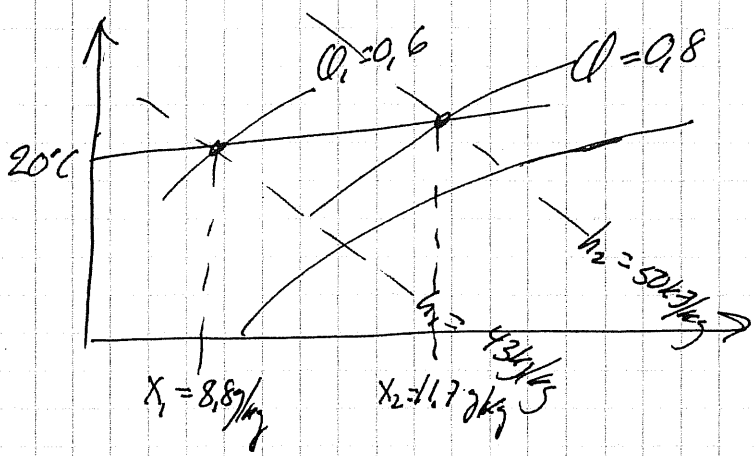
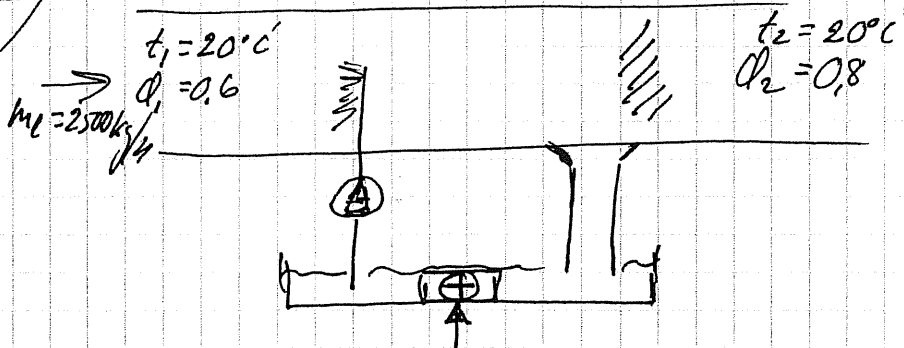
$$t_{\text{evap}} = \underline{15^\circ\text{C}}$$

$$\Delta X = 11.75 - 11.3 = 0.45 \text{ g/kg}$$

$$m_{\text{udsk}} = \frac{m_e \cdot \Delta X}{103} = \frac{3000 \cdot 0.45}{103} = \underline{1.35 \text{ kg/h}}$$

11.3 11.75 g/kg

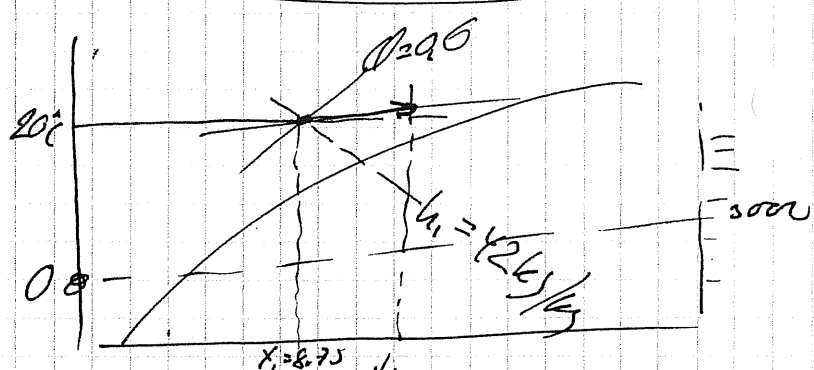
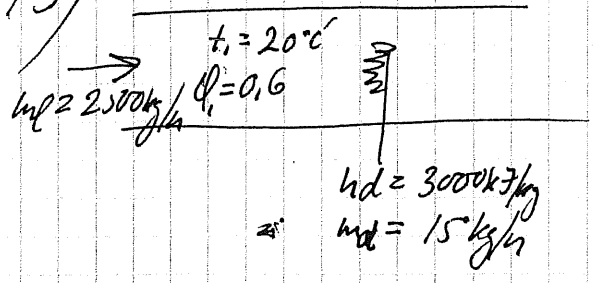
12)



a) opmærksom ved at holde vandet på en middeltemperatur 20°C. Altså effektivitet.

b) Tilført effekt. $P_{\text{eff}} = \frac{m_2}{3600} (h_2 - h_1) = \frac{2500}{3600} (50 - 43) = \underline{4.9 \text{ kW}}$

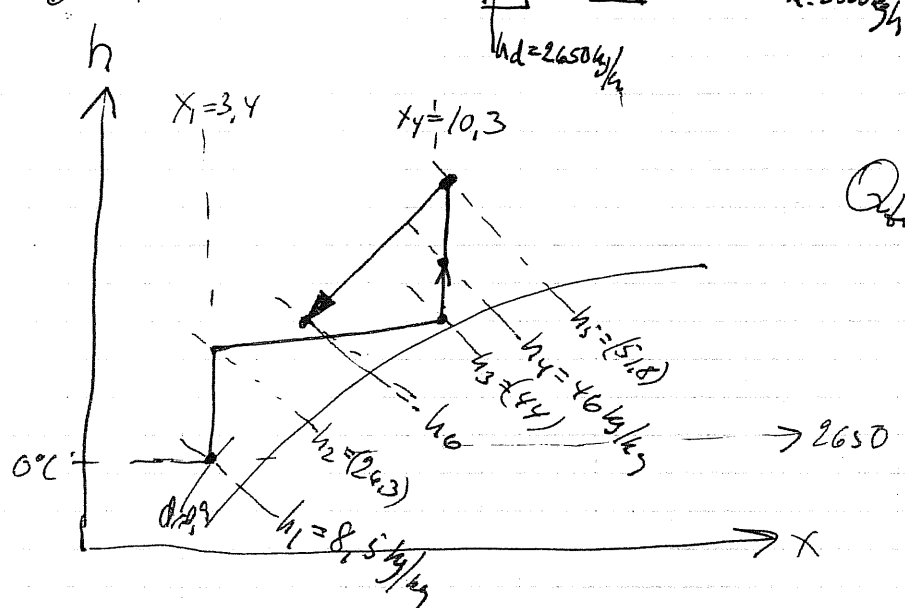
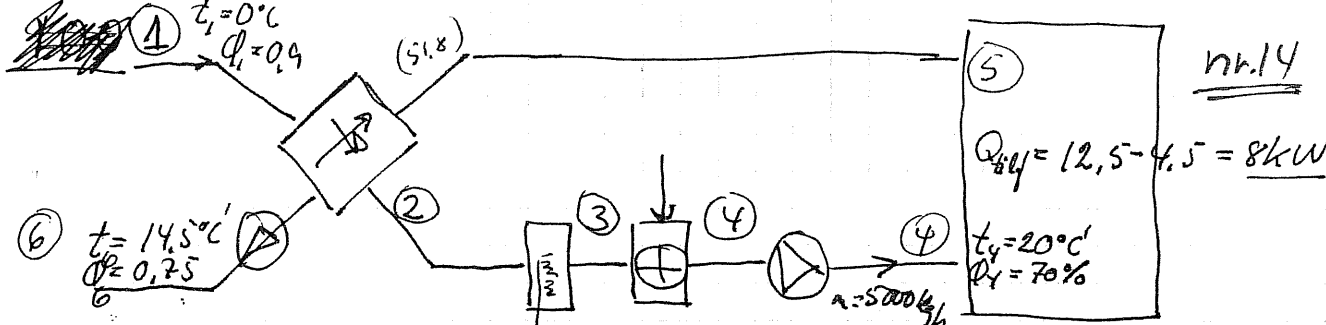
13)



$\Delta x = \frac{15 \cdot 10^3}{4000} = 3.75 \text{ g/kg} \Rightarrow x_2 = x_1 + \Delta x = 8.7 + 3.75 = \underline{11.45 \text{ g/kg}}$

$\Delta h = \Delta x \cdot h_d = \frac{3.75 \cdot 3000}{10^3} = 11.25 \text{ kJ/kg}$
 $\Rightarrow h_2 = h_1 + \Delta h = 42 + 11.25 = \underline{53.25 \text{ kJ/kg}}$

(tilstandene kan også bestemmes ved parallelforløb)



$Q_{\text{Lieferung}} = 12.5 - 4.5 = 8 \text{ kW}$

$\Delta h_{\text{trun}} = \frac{8 \cdot 3600}{5000} = 5.8 \text{ kJ/kg} \Rightarrow h_5 = h_4 + \Delta h = 46 + 5.8 = 51.8 \text{ kJ/kg}$

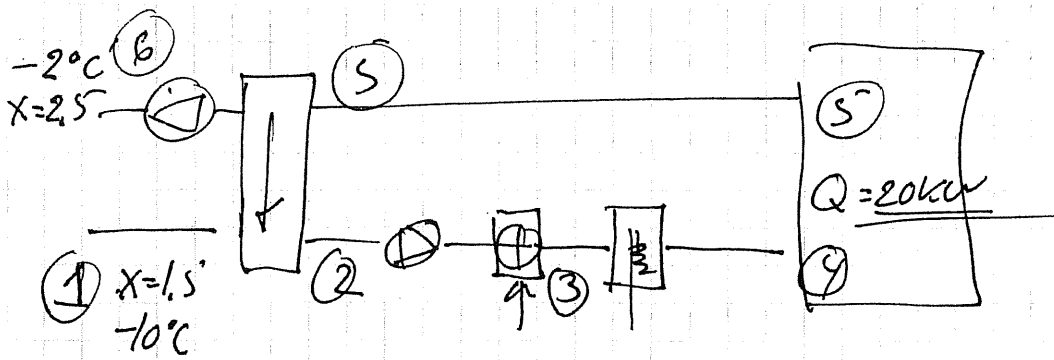
Ansatz entalpi i V.V = $\Delta h_{\text{trun}} = h_5 - h_6 = 51.8 - 34 = 17.8 \text{ kJ/kg}$

Ansatz entalpi = oplyst $\Rightarrow h_2 = h_1 + \Delta h_{\text{trun}} = 8.5 + 17.8 = 26.3 \text{ kJ/kg}$

Stampindspr. $\Rightarrow h_3 = 44 \text{ kJ/kg}$ $h_4 - h_3$

b) Effekt i effektflytt: $Q_{\text{eff}} = \frac{5000}{3600} \cdot (46 - 44) = 2.8 \text{ kW}$

a) Damp del beflytt $\dot{m}_v = \frac{5000}{1000} \cdot (10.3 - 3.4) = 34.5 \text{ kg/h}$



(Se selvst. h_x-diagram)

$\Delta X_{1 \rightarrow 2} = X_2 - X_1 = 5 - 1.5 = 3.5 \text{ g/kg}$ — sam overføres $\Rightarrow \Delta X_{5 \rightarrow 6} \Rightarrow X_5 = X_6 + \Delta X = 2.5 + 3.5 = 6 \text{ g/kg}$
 (temp 5 er opgjort til 22°C) \Rightarrow 5 er bestemt

Opvarmning fra 2 \rightarrow 3 med kasteret X \Rightarrow h₃ afleveres til 35 kJ/kg

Befugtn. ved kasteret er fulde p_i \Rightarrow h₄ = h₃ og punktet er fordelet
 ($h_{H_2O} = 0$)

fra 4 \rightarrow 5 er tør opr. i rum: ($\Delta h_{rum} = h_5 - h_4 = 37.5 - 35 = 2.5 \text{ kJ/kg}$)

$\dot{m} \Rightarrow \frac{\dot{Q}_{rum}}{\Delta h_{rum}} = \frac{20}{2.5} = 8 \text{ kg/s} \Rightarrow \underline{28800 \text{ kg/h}}$

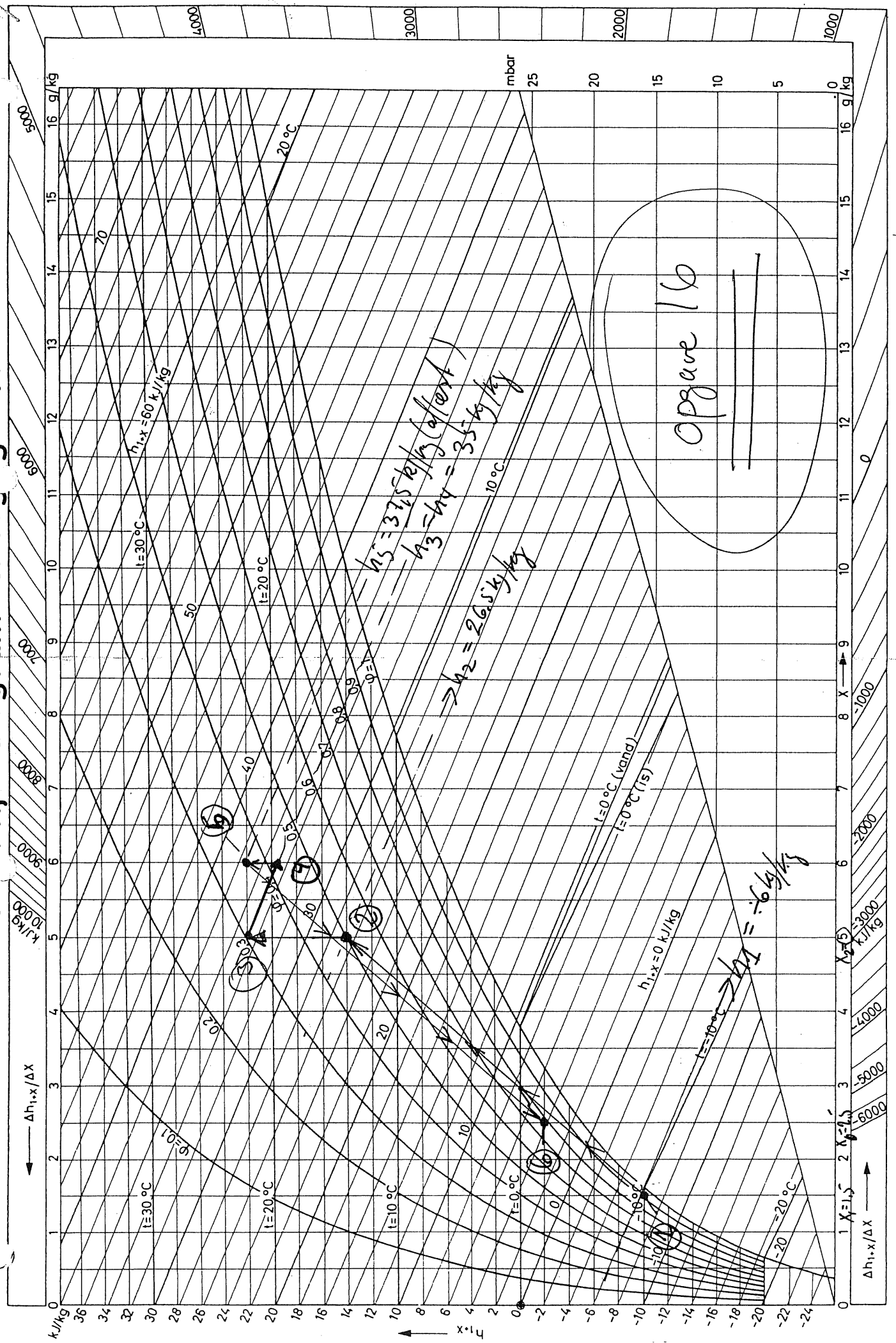
a) Q friskluft: $\dot{m} \cdot (h_2 - h_1) = 8 \cdot (26.5 - (-6)) = \underline{260 \text{ kW}}$

b) Vandmængde, friskluft opbej: $\dot{m} \cdot \frac{(X_2 - X_1)}{103} = 8 \cdot \frac{3600 \cdot (5 - 1.5)}{103} = \underline{100.8 \text{ kg/h}}$

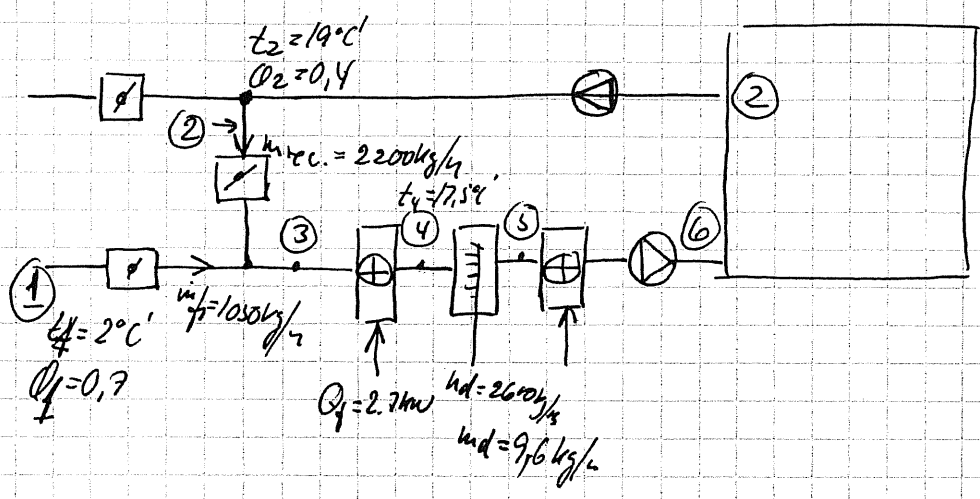
c) Vandmængde i befugter: $\dot{m}_{bef} = \dot{m} \cdot \frac{(X_4 - X_3)}{103} = \frac{8 \cdot 3600 \cdot (6 - 5)}{103} = \underline{28.8 \text{ kg/h}}$

d) Varmer effekt i V.V: $Q_v = \dot{m} \cdot (h_3 - h_2) = 8 \cdot (35 - 26.5) = \underline{56 \text{ kW}}$

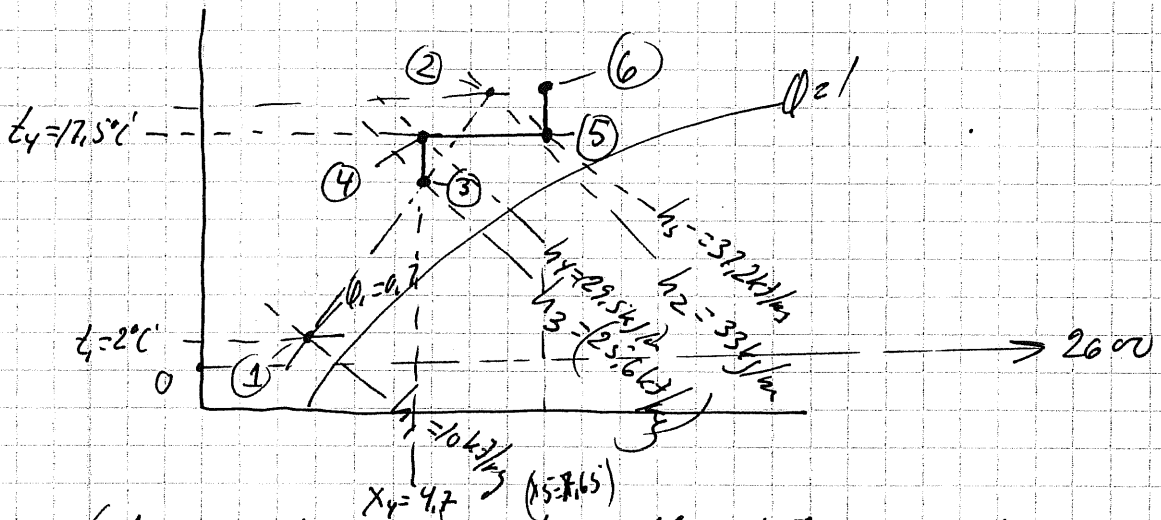
Molal ψ , η , χ -diagram for H_2O / H_2O



Opzave 16



Blandingspunkt $h_3 = \frac{m_{rec} \cdot h_2 + m_{pri} \cdot h_1}{(m_{rec} + m_{pri})} = \frac{2200 \cdot 33 + 1050 \cdot 10}{(1050 + 2200)} = 25,6 \text{ kJ/kg}$



($t_4 = 17,5^\circ\text{C}$ og $X_3 = X_4$) $\Rightarrow h_4$ aflæses til $29,5 \text{ kJ/kg}$

1) a) $Q_{forv} = (m_{rec} + m_{pri}) \frac{h_4 - h_3}{3600} = (1050 + 2200) \cdot \frac{(29,5 - 25,6)}{3600} = 3,5 \text{ kW}$

b) $\Delta X_{befugtev} = \frac{Q_{forv} \cdot 10^3}{(m_{rec} + m_{pri})} = \frac{3,5 \cdot 10^3}{(1050 + 2200)} = 2,95 \text{ g/kg}$

$X_5 = X_4 + \Delta X_{befugtev} = 4,7 + 2,95 = 7,65 \text{ g/kg}$

($\Delta h = \Delta X \cdot h_{d1} \Rightarrow h_5 = 29,5 + \frac{2,95}{10^3} \cdot 2600 = 37,2 \text{ kJ/kg}$)

ell. ved parallel forløbning

$\Delta h_{effervarefflet} = \frac{Q_{ev} \cdot 3600}{(m_{rec} + m_{pri})} = \frac{2,7 \cdot 3600}{(1050 + 2200)} = 3 \text{ kJ/kg} \Rightarrow h_6 = h_5 + \Delta h_{effervarefflet} = 37,2 + 3 = 40,2$

$\Rightarrow \left. \begin{matrix} t_6 = 21^\circ\text{C} \\ X_6 = 0,5 \end{matrix} \right\} \text{ aflæst}$

Mollier h,x - diagram for fuge 3 luft

